



Chemistry Practice test2

1. Formula making $10 \times 1 = 10$
 - a) Sodium nitrate
 - b) Sodium sulphite
 - c) Calcium bi carbonate
 - d) Copper sulphate penta hydrate
 - e) Nitric acid
 - f) Common salt
 - g) Potassium hydroxide
 - h) Ammonium chloride
 - i) Ferus chloride
 - j) Zinc sulphate
2. Give example of $5 \times 1 = 5$
 - a) Monovalent cation
 - b) Polyatomic anion
 - c) Diatomic element
 - d) Diatomic cation
 - e) Polyatomic element
3. Find atomicity of the following $5 \times 1 = 5$
 - a) $\text{Al}_2(\text{SO}_4)_3$
 - b) $2 \text{Na}_2\text{SO}_4$
 - c) K MnO_4
 - d) KCN
 - e) K_2O

4. Find molecular mass $5 \times 1 = 5$

- a) H_2
- b) H_2SO_4
- c) $10H_2O_2$
- d) $30 HNO_3$
- e) $20 H_2SO_4$

5. Name the followings $5 \times 1 = 5$

- a) $K_2Cr_2O_7$
- b) $KMnO_4$
- c) KCl
- d) KNO_3
- e) KOH

